

AD A105163

ANNUAL REPORT

LEVEL ¹¹

8
A091857

12

COMBUSTION MECHANISMS OF SOLID PROPELLANTS

By

E. W. Price, R. K. Sigman and R. R. Panyam

Prepared for

OFFICE OF NAVAL RESEARCH
ARLINGTON, VIRGINIA 22217

Under

Contract N00014-79-C-0674

September 1981

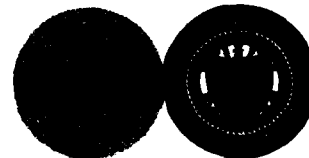
DTIC
ELECTE
S OCT 6 1981 D
D

GEORGIA INSTITUTE OF TECHNOLOGY

A UNIT OF THE UNIVERSITY SYSTEM OF GEORGIA
SCHOOL OF AEROSPACE ENGINEERING
ATLANTA, GEORGIA 30332

DISTRIBUTION STATEMENT A

Approved for public release;
Distribution Unlimited



81 10 5 074

FILE COPY

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER	2. GOVT ACCESSION NO. AD-A105143	3. RECIPIENT'S CATALOG NUMBER
4. TITLE (and Subtitle) COMBUSTION MECHANISMS OF SOLID PROPELLANTS		5. TYPE OF REPORT & PERIOD COVERED ANNUAL TECHNICAL REPORT, 1 Aug 1979-31 July 1980
7. AUTHOR(s) E. W. Price, R. K. Sigman and R. R. Panyam		8. CONTRACT OR GRANT NUMBER(s) ONR Contract No. N00014-79-C-0674
9. PERFORMING ORGANIZATION NAME AND ADDRESS GEORGIA INSTITUTE OF TECHNOLOGY SCHOOL OF AEROSPACE ENGINEERING ATLANTA, GEORGIA 30332		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS 12/371
11. CONTROLLING OFFICE NAME AND ADDRESS OFFICE OF NAVAL RESEARCH ARLINGTON, VIRGINIA 22217		12. REPORT DATE July 1981
14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office) POWER PROGRAM OFFICE OF NAVAL RESEARCH ARLINGTON, VIRGINIA 22217		13. NUMBER OF PAGES 28
		15. SECURITY CLASS. (of this report) UNCLASSIFIED
15a. DECLASSIFICATION/DOWNGRADING SCHEDULE		
16. DISTRIBUTION STATEMENT (of this Report) Approved for public release; distribution unlimited.		
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report) Approved for public release; distribution unlimited		
18. SUPPLEMENTARY NOTES		
19. KEY WORDS (Continue on reverse side if necessary and identify by block number) COMBUSTION, SOLID PROPELLANTS, ROCKETS, THEORY, SANDWICH BURNING		
20. ABSTRACT (Continue on reverse side if necessary and identify by block number) See next page.		

DD FORM 1 JAN 73 1473

EDITION OF 1 NOV 65 IS OBSOLETE

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

403914

501

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE(When Data Entered)

20. ABSTRACT

Investigation of mechanisms of combustion and combustion zone microstructure continued, using ammonium perchlorate-hydro-carbon binder sandwiches and the quench-burning method to obtain high resolution measurements. Binder-thickness deflagration limits were also determined, and work started on measurement of burning rate vs binder thickness. On the basis of results to date, a modified picture of the combustion zone is proposed that differs sharply from the classical model, particularly in those regions of the sandwich surface that are close enough to the AP-binder interface to correspond to processes in solid propellant combustion.

Accession For	
NTIS GRA&I	<input checked="checked" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
<input checked="checked" type="checkbox"/>	

DTIC
ELECT
OCT 6 1981
S D

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE(When Data Entered)

COMBUSTION MECHANISMS
OF SOLID PROPELLANTS

Prepared for

Office of Naval Research
Arlington, Virginia 22217

by

E. W. Price, R. K. Sigman and R. R. Panyam

School of Aerospace Engineering
Georgia Institute of Technology
Atlanta, Georgia 30332

Approved for public release; distribution unlimited.

ONR Contract N00014-79-C-0764

July 27, 1981

Conditions of Reproduction

Reproduction, translation, publication, use and disposal in whole or in part by
or for the United States Government is permitted.

ACKNOWLEDGEMENTS

The research reported was sponsored by the Power Branch, Office of Naval Research. Technical monitor of the project is Dr. R. S. Miller. The authors appreciate the support of ONR in this research, and the enlightened management of the project typical of ONR.

TABLE OF CONTENTS

Section	Page
Introduction	1
Experimental Method	2
Results	5
Burning Surface Profiles	5
Smooth Bands	7
Binder Thickness/Low Pressure Deflagration Limits	11
Theoretical Considerations	13
Single Interface Burning	14
Sandwich Burning	15
The Sandwich Flame Dimensions	17
Burning Surface Profiles	20
Deflagration Limits	21
Discussion and Summary	23
References	27

INTRODUCTION

Advances in prediction and control of combustion characteristics of heterogeneous propellants are severely limited by a lack of knowledge about structure and rate-controlling processes of the combustion zone. The marginal state of knowledge is due in large part to the microscopic dimensions and chaotic nature of the propellant and combustion zone structure. The chaotic microstructure has often been eliminated in past and present research by studies of propellant samples made from laminated sheets of oxidizer and binder, i.e., sandwiches. Real-time observations of edge-burning sandwiches can be made, with the location of the laminae, interfaces, and flame elements being established with fair accuracy. In addition, interrupted burning tests provide "post mortem" information about conditions in the condensed phase and burning surface to a much higher dimensional accuracy than observations during burning, because the quenched samples can be studied in optical and electron microscopes. From such studies, it has been possible to establish the combustion details on much less speculative grounds, and the emerging picture for propellants with ammonium perchlorate (AP) oxidizer and hydrocarbon (HC) binders is significantly different from the generally accepted models of combustion.

The present report summarizes studies of AP/HC binder sandwiches in the period 1 August 1979 - 31 July 1980, with some review of previous work to establish perspective. These current studies consisted of combustion photography and quench-burn tests on sandwiches with tapered binder laminae, i.e., tests in which the binder thickness was graduated from about 10 to 150 μ m (wedge). This approach was started in a previous program,^{1,2} and was continued here with emphasis on thin binder portions of the sandwiches (which are more relevant to propellants, and give combustion behavior different from behavior of the usual "thick-binder" sandwiches used in most other studies). The report describes quench profiles obtained over a range of binder thickness, pressure, and type of HC binder. Also reported are data on deflagration limits of sandwiches, and an heuristic argument relating observations to structure of the combustion zone.

EXPERIMENTAL METHOD

The experimental methods for preparing and testing sandwiches are described in a previous report (Ref. 1). Fig. 1 shows the sketches of tapered sandwiches. Fig. 1a shows a sample that has been burned down from a tapered edge, and Fig. 1b shows a sample that has burned down from the uniform thick edge. These particular examples were chosen because they illustrate the deflagration limit effect that occurs for thin binders. At pressures below the AP self deflagration limit, the sample in Fig. 1a burns down until quenched by rapid depressurization of the test vessel, but a portion of the "ignited" edge where the binder thickness is $< 40 \mu\text{m}$ fails to burn. The sample in Fig. 1b burned down uniformly from the ignited thick-binder edge until the binder had decreased to $40 \mu\text{m}$ or less, and then went out spontaneously. The tests illustrated by Fig. 1a were used to obtain details of the surface profile vs binder thickness, while tests illustrated by Fig. 1b were used to determine deflagration limits. Details of the tests are summarized as follows:

All sandwiches were made in the same manner, by dry-pressing granular AP into 1.27 mm thick sheets and curing a binder lamina in place between two sheets. The AP used was Kerr-McGee 99.7% pure ground and screened to unimodal particle size in the 75 to 120 μm size range; pressing was at 165 MPa for 60 min. Binder ingredients were conventional propellant grade, provided by the Naval Weapons Center (PS, PBAN, HTPB, CTPB). Binder thickness was controlled by using a spacer between the AP sheets on one edge of the sample, and no spacer on the opposite edge. Actual thickness was determined from the quenched samples rather than from control of initial sample dimensions.

In tests, the samples were mounted in a nitrogen-pressurized chamber with vent at the top. The samples were ignited on the top edge with a hot wire and igniter paste. When controlled sample quenching was desired, it was accomplished by rapid depressurization ($\sim 3 \times 10^3 \text{ MPa/sec}$) due to blow-out of a plastic burst diaphragm. Samples were subsequently coated with an evaporated film of gold, and examined in a scanning electron microscope (SEM). In the case of deflagration limit tests, several measurements were made of binder thickness along the quenched surface, using pictures from the SEM. In tests with combustion photography, a conventional nitrogen-flushed combustion chamber was used, and 16 mm motion pictures were taken at 2200

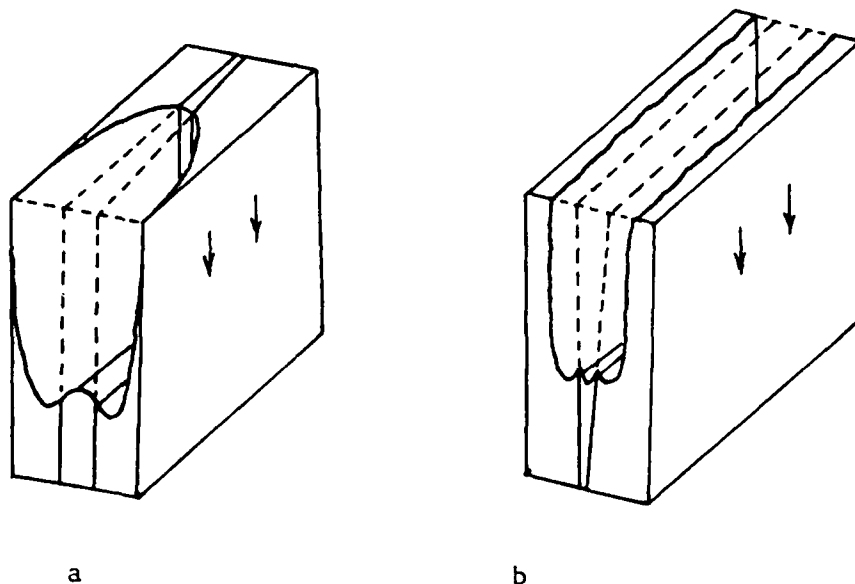


Fig. 1 Quenched tapered binder sandwiches:
a) Sandwich burned on a tapered edge at 2 MPa and quenched by rapid depressurization.
b) Sandwich burned from the thick edge at 2 MPa and allowed to quench spontaneously.

frames per second with a picture width corresponding to 1 cm of the object plane.

Exploratory tests were also made to determine the possibility of obtaining burning rate in sandwich tests. This is difficult at most rocket motor pressures because the recessed nature of the leading edge of the burning front prevents accurate position determination from photographs during burning. Preliminary tests were run on "wired" sandwiches (Fig. 2) which would indicate passage of the burning front at intervals during burning. No successful tests were achieved, but work is continuing.

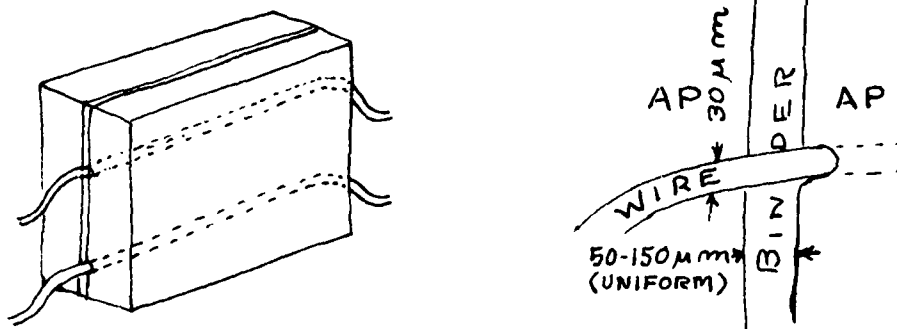


Fig. 2 Wired sandwiches for burning rate measurements.

RESULTS

As indicated above, the tests results consist primarily of quenched samples, which are studied by microscopic methods to determine surface profiles, surface details and binder thickness. The relation of these results to test conditions, and to general understanding of combustion mechanisms provide the basis for a qualitative theory of combustion presented in the Discussion.

Burning Surface Profiles

Observation of surface profiles was primarily a confirmation of results observed previously,^{1,2} and reaffirmed that profiles are critically dependent on binder thickness. A sketch of profile trends is shown in Fig. 3. In anticipation of later discussion, it is relevant to note both classical and newly recognized features of profiles.

The most universally reported feature of surface profiles is the trend with pressure. Below the AP deflagration limit (parts a and d of Fig. 3), the

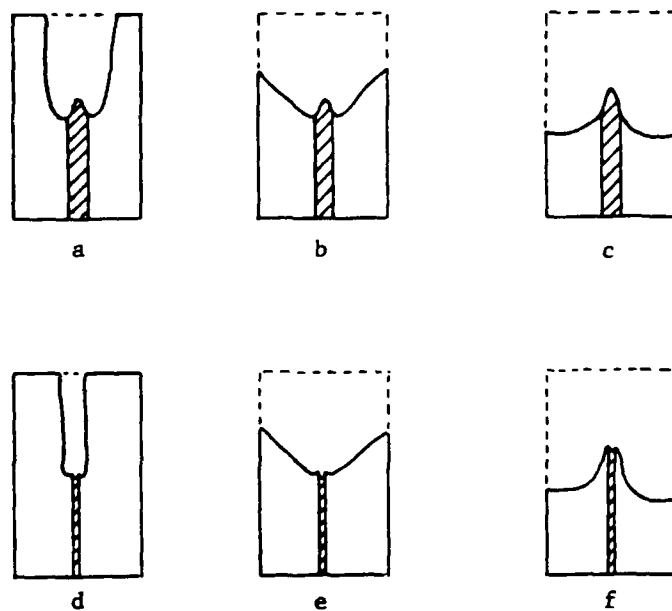


Fig. 3 Cross-section profiles of sandwich burning surfaces (sketched from quenched samples):
a) Low pressure (~ 2 MPa); b) Intermediate pressure (~ 5 MPa); c) High pressure (~ 8 MPa); d, e, f Same as a, b and c, but thin binder ($\sim 50 \mu\text{m}$).

burning proceeds down the binder lamina, consuming an amount of AP roughly commensurate with a stoichiometric ratio to the binder. At intermediate pressure (parts b and c of Fig. 3), the AP deflagrates at its own characteristic rate, with the profile piloted by a faster-burning interfacial region at the "sandwich" burning rate (when the characteristic AP burning rate is known, the slopes of the "V" profile can be used to calculate the sandwich rate). At high pressure the AP deflagration rate exceeds the sandwich rate, with the result that the binder lamina protrudes above a flat AP surface (parts e and f of Fig. 3).

A more recently established feature of sandwich profiles is the trend with binder thickness. A result noted in limited tests in Ref. 3 - 5, and many tests in Ref. 1, 2 is the change from protruding to recessed binder surfaces as binder thickness is reduced (Fig. 3 a - c vs Fig. 3. d - f). This trend has been consistently supported in current tests as well, although there is some quantitative difference with different binders. These trends are currently being studied with families of sandwiches with uniform thickness, to determine consistency of results at each binder thickness.

A relatively recent^{1,6} (and ongoing) observation about surface profiles concerns the details of the AP profile near the AP-binder interfaces. Early papers reported the leading edge to be at the interface, but the work reported in Ref. 3 showed that the leading edge of the AP was 20 - 100 μ m out from the interfaces as implied in Fig. 3. Subsequent studies using sufficiently high resolution methods consistently confirmed this result and extended it to a wide range of test conditions (Ref. 1,2,4-6). In some instances it was obvious that the effect resulted from retardation of the AP regression rate near the interfaces by a flow of molten binder onto the AP.^{3,7,8} It is only in recent work^{1,2,6} that it became clear that the retardation of the AP regression rate is present even in the absence of binder melt flow, and with all binder thicknesses at all pressures. This has been established in the present studies for PBAN, PS, and HTPB binders, for binder thicknesses from 20 - 150 μ m, and pressures from 13.8 MPa down to the sandwich low pressure deflagration limit. This point will be illustrated in the next subsection in connection with results concerning other attributes of this region of the burning surface.

Smooth Bands

In the previous report^{1,2} it was noted that the surface quality of the quenched AP laminae in sandwiches (and propellants) was different near the AP-binder interface than elsewhere on the samples. While the majority of the AP surface exhibited the surface undulations and froth residue typical of AP self deflagration, the region near the interface plane exhibited an almost "glossy" appearance in scanning electron microscope pictures (Fig. 4). Current work has confirmed that this "smooth band" adjoining the interface plane is coincident with the region of retarded regression of the AP surface (Fig. 5), indicating that these are two separate manifestations of a single attribute of the local AP deflagration that is different from that elsewhere on the sample, a difference linked to the proximity of the binder lamina.

The possibility that the smooth band effect is a result of binder melt flow seems to be eliminated because in results to date the band is persistently present, even in situations not conducive to flow. These situations include: relatively dry-burning binders such as polysulfide; thin binders that burn in a recessed manner (Fig. 5,6); samples burned upside down so the gravitational effect resists flow onto the AP surface. Smooth bands and locally retarded AP regression occur in all these situations.

Another possibility that was considered as a possible cause of smooth bands and retarded regression was that, during sandwich fabrication, some binder or fuel-like ingredient might diffuse into the AP lamina to give a modified layer with modified deflagration characteristics. This possibility was rejected on several grounds. There was no direct evidence of fuel diffusion. If it did occur, it was reasoned that it would increase regression rate, not retard it. Further, the same effects (smooth band, retarded rates) seem to have occurred consistently also in earlier testing³⁻⁵ where the AP laminae were cut from large single crystals that would be nonporous. In addition, tests were run in the present study using AP laminae pressed to lower density, and laminae pressed from very fine (5 μ m) AP. These modifications had no noticeable effect on the smooth band phenomenon. Thus it is concluded that fuel permeation is not responsible for the smoothness and rate effects.

An earlier critical test regarding smooth bands involved interruption of burning of a sandwich in which the fuel lamina was replaced by a sheet of



Fig. 4 Scanning electron micrographs of quenched tapered sandwiches showing characteristic surface profiles and surface details such as smooth bands. Binder thickness in these pictures ranges from $70 \mu m$ (left) to $10 \mu m$ (right).

4a AP/CTPB/AP sandwich quenched from 1.38 MPa. Top shows most of sample, lower shows smooth band, with recessed binder at top.

4b AP/PS/AP sandwich quenched from 4.1 MPa. Top shows the thin binder end of the burning surface, lower shows smooth bands on each side of binder, typical AP self-deflagration surface further from binder. (PS binder gives little or no evidence of melt or flow.)

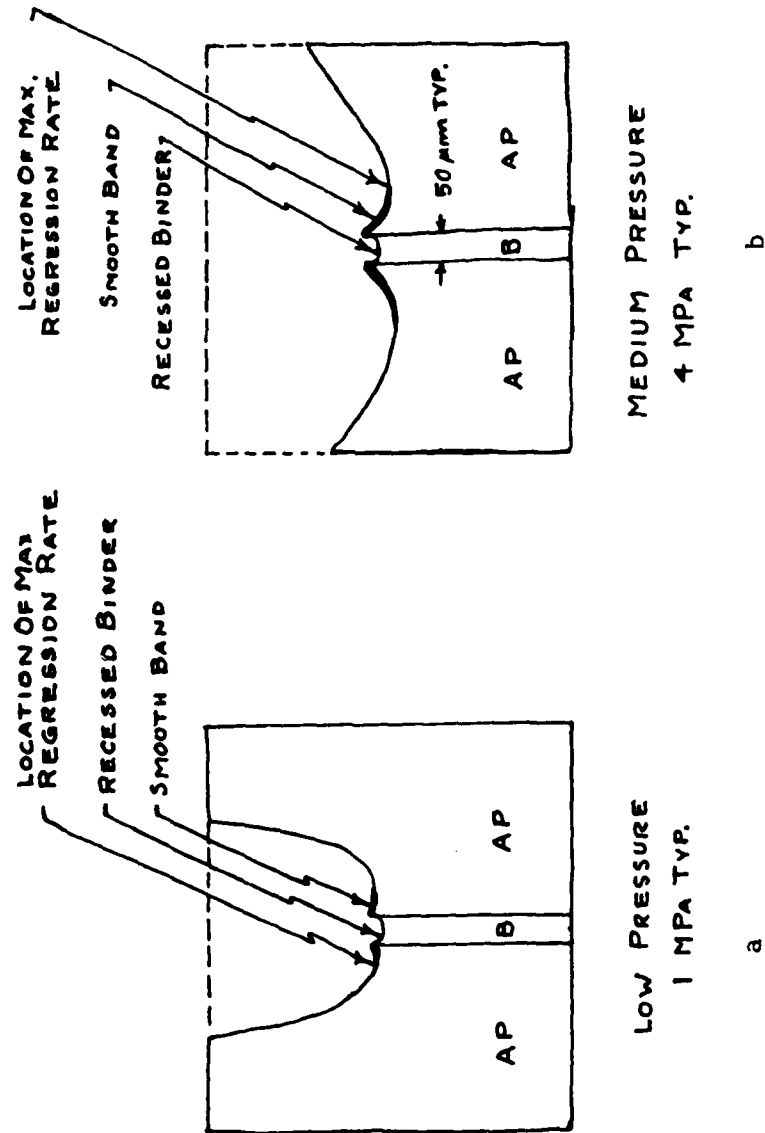


Fig. 5 Sketches of profiles of thin-binder sandwich sections, showing locations of characteristic features of profiles and burning surfaces. 5a Typical of 1 MPa; 5b Typical of 4 MPa.

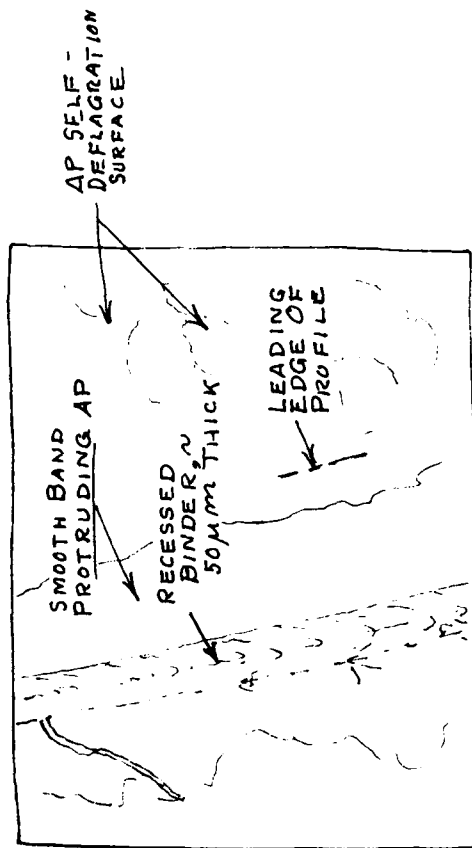


Fig. 6 Uniform-binder sandwich quenched from 4.14 MPa (AP/PBAN/AP, binder thickness 30 μ m). Note retardation of AP regression near the binder interface, and the relatively smooth quality of the surface in this region. Note region of maximum AP regression is near "outer edge" of smooth band.

of mica.^{1,2} This test resulted in absence of the smooth band and of the retarded rate effect. This test was repeated in the present study with the same result. Mica was chosen originally because it is a non-reactive thermal insulator. Its use removes the AP-binder flame and minimizes heat transfer in the non-AP lamina. Tests are in progress on sandwiches with non-reactive, thermally conductive laminae (gold leaf), to try to determine to what extent the "mica results" are due to removal of the AP-binder flame and to what extent they are due to changes in heat conduction. Preliminary results indicate that the smooth band and retarded AP regression are present when a 14 μm thick gold lamina is used.

Binder Thickness/Low Pressure Deflagration Limits

Tests such as that illustrated by Fig. 1b yielded a set of measurements of binder thickness from each spontaneously quenched sample, an average value, and a range of values. These results are shown in Fig. 7 as a graph of binder thickness vs test pressure, for three different binders. The results indicate the following features:

1. There is a minimum binder thickness for self deflagration of sandwiches, which is relatively insensitive to pressure down to about 0.4 MPa. This minimum value of thickness is about 20 μm for PBAN binder, and 30 μm for PS and HTPB binder.
2. Below about 0.4 MPa, the limiting thickness of binder increases rapidly with decreasing pressure.
3. Self quenches occurred in some tests at pressures above the stated AP self deflagration limit.
4. The range of binder thickness in any one test sample was fairly large, especially with PS binder. This data scatter, which is common to most flammability measurements in combustion science, seems nonetheless to establish items 1 - 3 above beyond doubt for this experiment.

There also appears to be a significant but small systematic trend in binder lamina thickness vs pressure in the > 0.4 MPa range which may also prove useful in mechanistic interpretations, but reproducibility probably does not warrant speculation now.

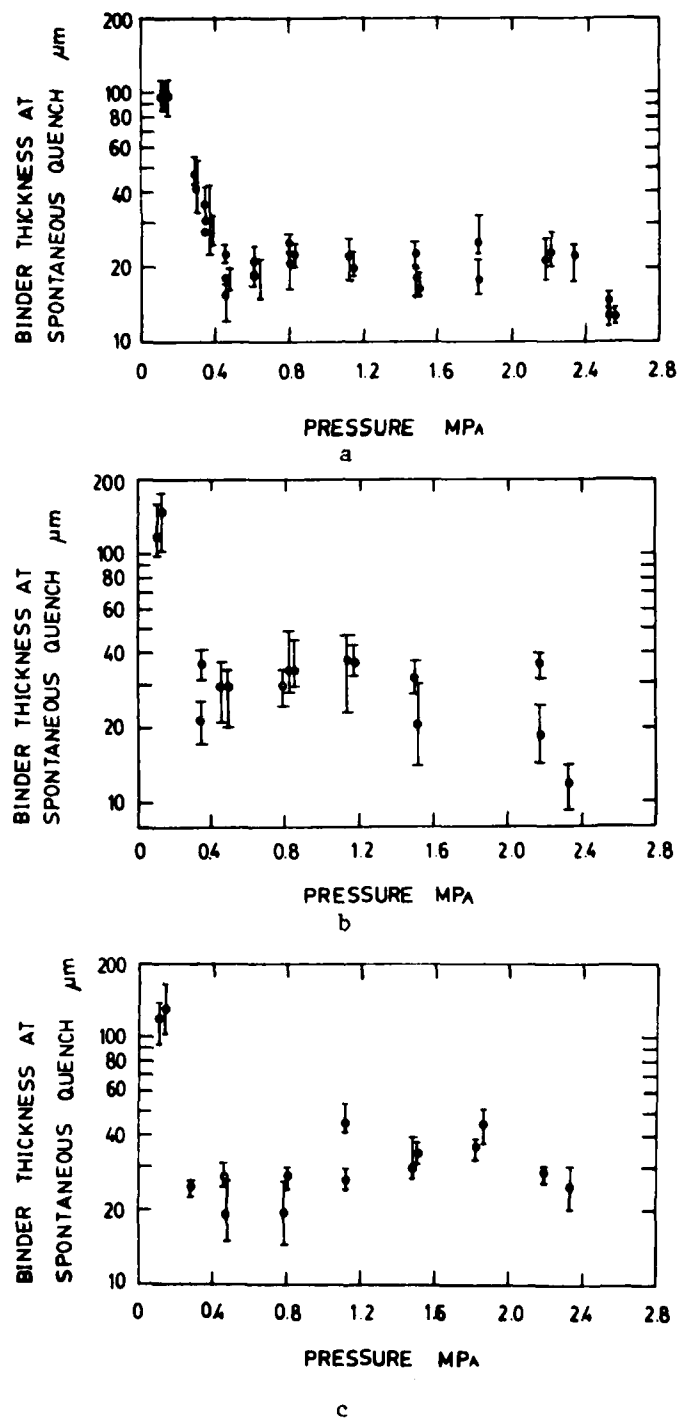


Fig. 7 Deflagration limits for tapered binder sandwiches burned at constant pressure and spontaneously quenching at the indicated binder thickness: a) PBAN binder; b) PS binder; c) HTPB binder.

THEORETICAL CONSIDERATIONS

Ideally, one would like to use the experimental results for guidance in formulation of an analytical model of sandwich burning, and then test the detailed agreement between model and experiment. In this, one would be guided by previous efforts and hope for refinement of those results. However, it is doubtful that an analytical model of AP-HC sandwich burning can be constructed now that is both realistic and tractable. That would seem to be a longer range goal, that is one of the mileposts on the trip to propellant modeling. On the other hand, there is a great deal known about the combustion mechanisms, and it appears logical to construct a qualitative description that can serve as a basis for interpretation of current experimental results, for design of future experiments, and for constructing analytical models.

The more generally accepted views of sandwich burning mechanisms are embodied in Fig. 8. The oxidizer (AP) burns with its own flame, standing close to the surface. In addition, there is substantial support for the view that an exothermally reacting melt layer is present at pressures > 2 MPa, i.e., above

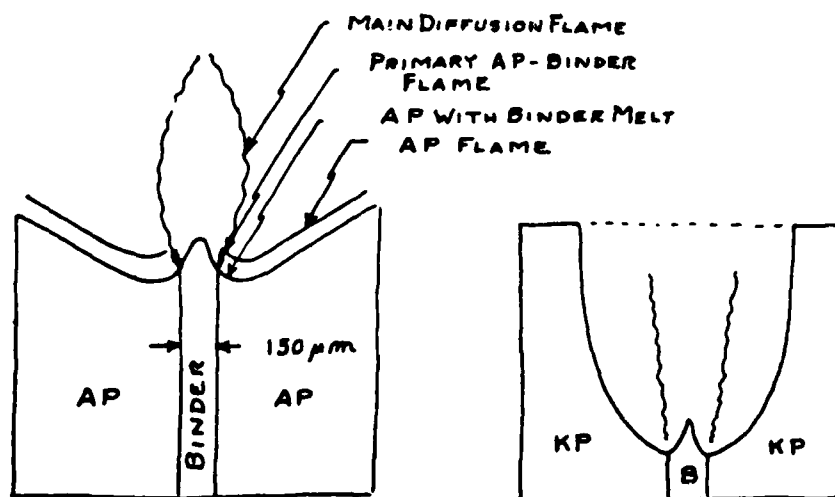


Fig. 8 Conventional view of the combustion zones of AP/HC, and KP/HC sandwiches.

the AP low pressure deflagration limit. Relatively little is known about the AP flame below 2 MPa, although it is plausible to assume that it is present at lower pressures in regions where the AP-binder flame supplies sufficient heat. When the flame is absent, the AP probably decomposes endothermally by dissociative sublimation to NH_3 and HClO_4 .

In the micro region close to the AP-binder interface, the oxidizer vapors and binder vapors mix quickly, to give the kinetically controlled "phalanx" flame proposed by Fenn.⁹ There is no direct experimental evidence of the location or influence of the phalanx flame. In some respects it is analogous to the "primary" flame used by Beckstead¹⁰ and others in modeling combustion of propellants. However the "primary" flame is defined arbitrarily as the portion of the oxidizer-fuel (O-F) flame closer to the surface than the AP flame. It is not clear whether this is more, or less, than the kinetically limited phalanx flame of Fenn (who was not considering a self deflagrating oxidizer). In any case, a portion of the O-F flame more distant from the surface is diffusion-limited, because of the longer diffusion paths for reactants at those locations. As noted by Summerfield,¹¹ the portion of the flame that is kinetically limited increases as pressure is reduced, because kinetic rates decrease more rapidly with pressure than diffusion rates do. In the case of AP sandwiches, the pressure dependence of the overall flame complex is further enhanced at low pressure by the tenuous nature of the AP flame.

The sketch in Fig. 8 shows a protruding binder lamina, and an AP profile that slopes up to a protruding binder. As noted earlier, the protrusion of the AP has been recognized for many years. However the reason for the protrusion has not been established, and only recently has it been recognized that the condition exists even when the binder is recessed. This question will be addressed later in this report, but to do so it is advantageous to go back and address first some more general questions about how sandwiches burn. This will be done by starting with the simpler situation, the burning of sandwiches with "non-selfdeflagrating" oxidizers. This is the situation assumed in most analytical models.^{9,12-14}

Single Interface Burning

Consider first the burning of a "composite" consisting of a slab of binder

against a slab of non-self deflagrating oxidizer, ignited at an exposed edge of the interface plane. The flame is pictured as in Fig. 9, consisting of a leading edge that is a kinetically limited (pre-mixed) flame, with a transition to a trailing diffusion-controlled flame further from the interface. Both analytical and experimental evidence of the structure of this flame complex is speculative, involving as it does microscopic dimension and uncertain information about transport and kinetic properties in the region. For convenience, a region of the leading edge of the flame complex is identified as a "propagation velocity controlling" (PVC) region, i.e., that part of the flame that contributes to pyrolysis of the solids at the leading edge of the surface profile. Because it is difficult to model the PVC region accurately, it is also difficult to explain the surface profile accurately—but accurate experimentally determined profiles may provide good tests of the validity of models. For the present purposes, the classical assumption will be adopted, that the PVC region includes both the premixed flame and part of the diffusion flame, with the relative contributions from the two parts being dependent on pressure, etc.. The contribution of an AP decomposition flame, if any, will be considered later. Thus single interface burning is described as a PVC flame that determines the leading edge velocity and profile, with a trailing flame that sustains burning of the ingredients further out from the interface.

Sandwich Burning

Most experimental work on burning of laminated systems has been done on sandwiches, consisting of a fuel lamina between two laminae of oxidizer. If the fuel lamina is thick enough, such a system will burn with two substantially independent interface flame fronts as in Fig. 10. Such flame systems would presumably interact with each other only at the trailing edge of the lamina. Most investigators have chosen fuel laminae that are thin, in the sense that the overall sample stoichiometry is fuel lean. Under these conditions the burning proceeds down between residual walls of oxidizer, with the fuel lamina protruding moderately from the bottom of the slot. The extent of fuel lamina protrusion, and hence of separation of the PVC regions, varies with the fuel lamina thickness (Fig. 11), as well as fuel pyrolysis characteristics, pressure, etc.

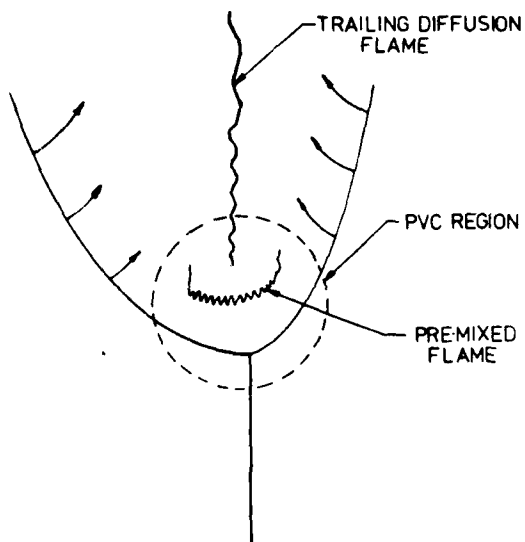


Fig. 9 Model of burning down the interface between an oxidizer and a fuel slab.

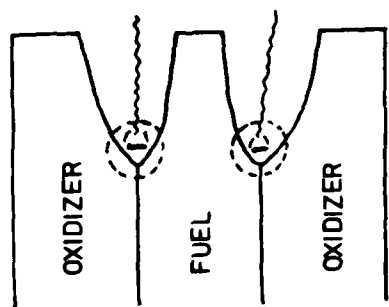
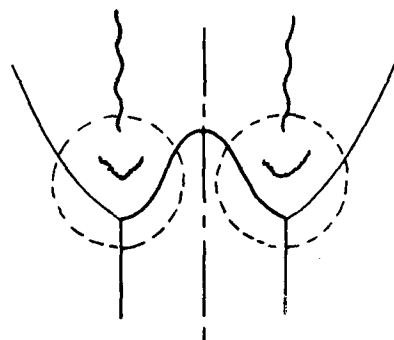
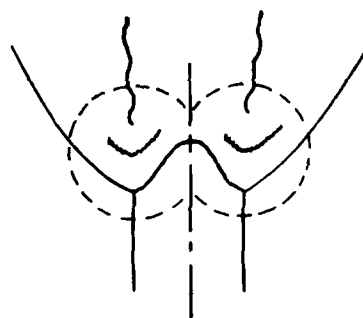


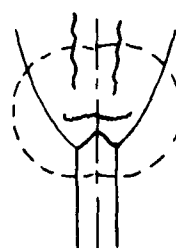
Fig. 10 Combustion of a sandwich with thick fuel layer (two non-interacting PVC flames).



a



b



c

Fig. 11 Effect of reducing fuel lamina thickness (showing onset of PVC region interaction): a) intermediate thickness; b) thin binder; c) very thin binder.

In order that the results of studies on sandwiches will be relevant to combustion of heterogeneous solid propellants, one must examine what aspects of sandwich burning can in fact approximate some aspect of propellant combustion. Evidently, the first requirement is that behavior be simulated and examined on the same microscopic dimensional scale as the propellant combustion zone microstructure. It apparently does not matter whether the sandwich is fuel lean or fuel rich, so long as that does not affect the PVC region.* But if the binder lamina is too thick, the interaction of PVC region pairs may be prevented, or excessive binder melt may flow into the PVC region. Since the actual dimensions of the PVC regions are not known, their dimensions will be as much a goal of experiments as a guide to design of them. At the outset, the characteristic dimensions of the propellant microstructure suggest that the spatial domains of relevance in the solid are from 0 to about 200 μ m in the oxidizer, with typical thickness of binder structural elements being from 0 to 50 μ m. Consequently the present studies have addressed primarily sandwich burning results with binder thickness in the 0 - 70 μ m range with larger thickness being used primarily to provide continuity with that large portion of previous sandwich burning research that used binder thicknesses > 70 μ m.

The foregoing strategy for relevance does not settle the question of dimensions of the PVC region in a fundamental sense; it only sets limits on dimensions of the part of the PVC region that actually have an opportunity to occur in the propellant combustion zone. This is closely related to the issue of proper modeling of propellant combustion and the effect of particle size on propellant combustion. The issue will be examined here in the restricted scope of the sandwich combustion zone.

The Sandwich Flame Dimensions

When the binder lamina is fairly thick, diffusion processes lead to heat and oxidizer diffusion** laterally out of the PVC regions into a region of the binder lamina and to fuel flow that ultimately is convected away between the

* A somewhat larger region may be involved in other aspects of burning such as oscillatory combustion.

** Heat diffusion via both solid and gas phase.

PVC regions (Fig. 11a). As the binder lamina is taken thinner, its tip retracts (Fig. 11b, c), while the PVC regions come closer together, and the loss of heat and oxidizer from between the PVC regions decreases until the PVC regions merge into a single combustion wave (Fig. 11c). One might anticipate that this more conservative flame would burn more rapidly into the sandwich, and that the binder thickness where the increase in burning rate showed up would be an indicator of dimensions of the PVC region. While there is only minimal data available on burning rate of sandwiches, Fig. 12 is a rough sketch of trends reported in Ref. 12-17, showing burning rate as a function of binder thickness for potassium perchlorate and ammonium perchlorate systems, adjusted to a pressure of 2.0 MPa. As the binder thickness is reduced from a large value, a burning rate increase begins at about 1 - 2 mm for KP. A similar but less conspicuous effect is observed for AP sandwiches, with the binder thickness being 0.5 - 1.0 mm. Thus the two PVC regions begin to affect each other when interfaces are 0.5 - 2 mm apart, implying a fuel-side width of the PVC region of a single interface flame of about 250 - 1000 μ m (pressure dependent).

The broken-line parts of the burning rate curves in Fig. 12 were inserted without any direct experimental data, to indicate an expected trend to decreasing burning rate with decreasing binder thickness. This trend is due to increasing lateral losses on the oxidizer sides of the flames as the flames burn down an increasingly narrow slot in the sample. In this thickness range the two flames are presumably fully merged on the fuel sides, so that no compensating "gains" occur on the fuel side to keep the burning rate up as fuel thickness is reduced, so the exterior loss trend dominates. The validity of this postulated trend of burning rate with binder thickness is supported not only by the above reasoning, but also by extrapolation of experimental results of Ref. 14 and by the present observation of minimum thickness for sustained burning of sandwiches (Fig. 7). The thickness scale of the curves in Fig. 12 was chosen on the basis of reconciliation of experimental sources (Ref. 12-17), with the dominance of outer "losses" over inner "gains" occurring when binder thickness is reduced below 100 μ m (AP) - 300 μ m (KP).

The trend of the thin-binder ends of the curves in Fig. 12 terminates at the deflagration limit. No limits were found in the literature for KP sandwiches, but those for AP are reported here in Fig. 7. Tests on KP sandwiches with 125 μ m PBAN binder were attempted, but ignition could not

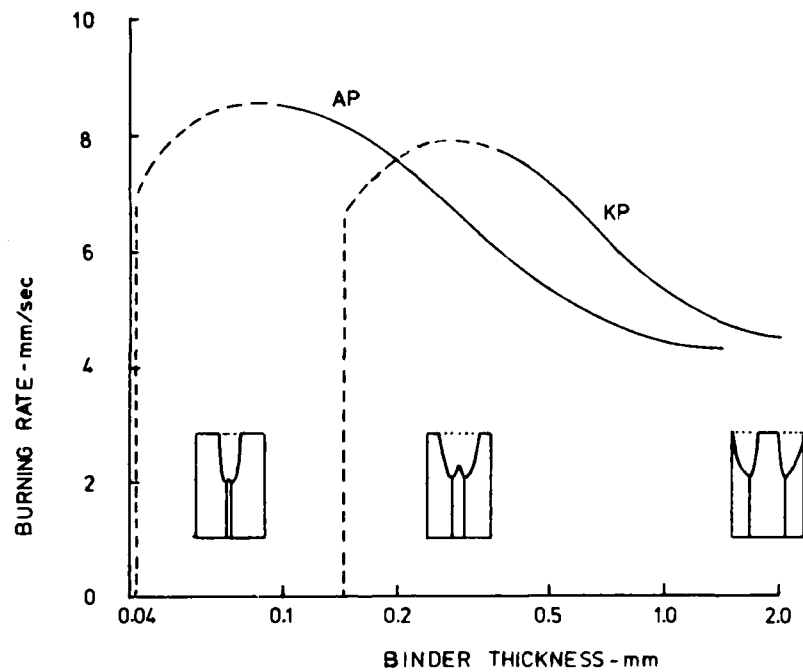


Fig. 12 Trend of sandwich burning rate with binder thickness.

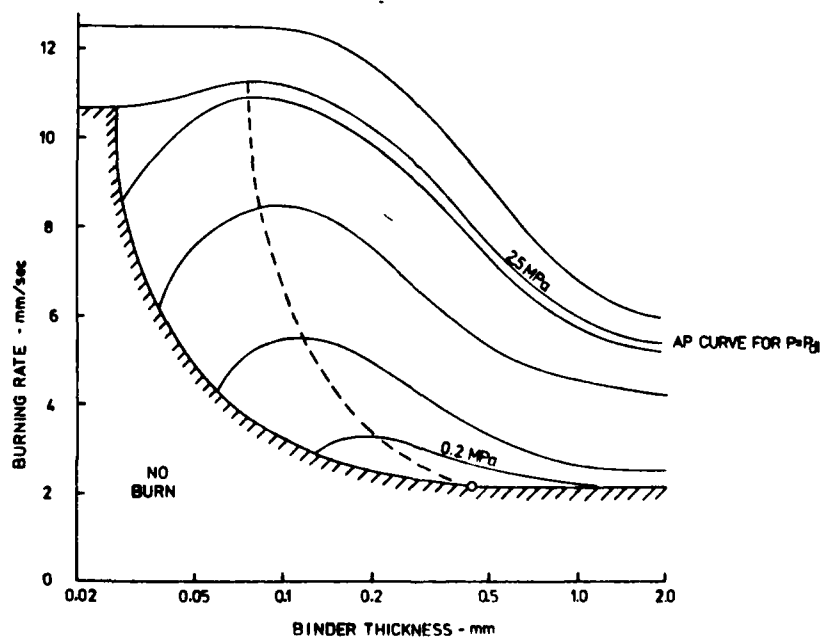


Fig. 13 Trend of sandwich burning with binder thickness and pressure, for oxidizer that self deflagrates above 2.5 MPa.

be achieved at any pressures tested (2.2 MPa to 6.9 MPa). It should be noted that the curves in Fig. 12 are only estimated. The results in the literature represent a variety of experiments, often only superficially described, and with results that either should not, or could not, be quantitatively reconciled. However, they do seem to provide a reasonable basis for the qualitative trends described here. In combination with the interpretation of the results, a basis is provided for estimates of the dimensions of the PVC regions and the burning surface profiles (sketched in Fig. 12 and discussed shortly).

The foregoing can (albeit recklessly!) be extended to a general picture of sandwich burning as in Fig. 13. In this figure, the more gross contributions of AP self-deflagration (neglected to this point) are introduced as a change in burning rate trends with pressure, and a change of the AP sandwich deflagration limit to zero thickness above 2.5 MPa. Obviously, the argument needs to be tested by some systematic burning rate measurements, but present results suggest a PVC region that is between 25 and 100 μ m wide on the fuel side, with a similar stand-off distance from the surface.

Burning Surface Profiles

To date, no realistic analytical models of sandwich burning have been developed, and even the simpler process of single interface burning suffers from lack of complete description of the PVC region, even for KP oxidizer. Thus most of the generalizations reached in the present work regarding surface profiles can be discussed in only qualitative terms. The big uncertainty seems to be the self-deflagration behavior of the oxidizer. When the AP deflagrates, it produces a flame close to the burning surface, that is believed to contribute appreciably to heat conduction to the burning surface. But under what conditions does that flame occur? Does it occur in the PVC region? Does the diffusion of fuel species enhance or suppress the AP flame? How does the lateral heat flow in the laminae affect the AP flame? None of these questions can be answered decisively at present, yet answers would be necessary for formulation of a combustion model.

There was one striking contrast between the present results with AP sandwiches and those predicted and observed with single interface models and KP-fuel slab tests, i.e., the retardation of the oxidizer regression adjacent to

the interfaces. At high pressure (e.g., 10 - 14 MPa) this can be regarded as a case of dominance of the AP self-deflagrating rate over the interface burning rate. This interpretation is not likely to explain the extreme protrusion of both binder and AP that actually occurs (Fig. 3c, 3f, 6). It is postulated that the extreme protrusion is due to failure of the AP flame to maintain itself near the interface. This hypothesis is supported not only by the profiles, but also by independent evidence that the AP self-deflagration flame is only marginally stable in the absence of fuel lamina,¹⁸⁻¹⁹ with the low pressure deflagration limit being very sensitive to impurities, heat losses, and ambient sample temperature. In sandwich burning, the AP near the interface loses heat to the endothermic binder, an effect that would quench the AP flame, with corresponding reduction of heat transfer rate to the surface, reduction of regression rate, and a corresponding protrusion of the interface AP.

At low pressure, one may call on substantially the same argument. One may argue that the AP flame persists in favored locations even below the low pressure deflagration limit provided it is supported²⁰ by the AP-fuel flame. However, the AP flame would, as argued for the high pressure situation, fail to sustain in close proximity to the interfaces where losses to the endothermic binder lamina are excessive. This would explain the continued retardation of interface AP, even at low pressure. It may also explain the most striking observation of the present tests, the correspondence of the locations of the smooth band and the region of retarded AP regression adjoining the interfaces. We suggest that the smooth band is associated with the absence of an AP flame.

These rather speculative interpretations of persistent effects in the sandwich tests have fairly profound implications for propellant combustion, where the entire burning surface is made up of structural entities of the same dimensional order as smooth bands, retarded regression profiles and postulated AP flame quenched regions. In the meantime, the collected results give a clue as to flame zone structure and dimensions.

Deflagration Limits

It has been noted that the thin-binder sandwiches are very fuel lean, and below the AP deflagration limit the sandwiches burn down a narrow slot in the

sample. In burning down tapered sandwiches, the PVC flame is subject to increasingly high (proportional) heat loss as the binder thickness decreases and the slot becomes more narrow. The resulting reduction in PVC flame temperature would be expected to lead to reduced reaction rates, and ultimately to quenching. In the deflagration limit experiments, this situation is reached in each test. At pressures about 0.4 MPa, the quench occurs at a binder thickness that is relatively insensitive to pressure (Fig. 7). This result is interpreted here as implying that diffusion, rather than kinetics, is dominant in controlling either reaction rate in general, or the shape of the flame. The relative pressure independence of diffusion controlled burning then controls the approach to the deflagration limit. Kinetic considerations presumably become more dominant near the deflagration limit, due to the decrease in flame temperature in the narrowing slot. If this interpretation is correct, there is no way to determine from these results whether the moderate dependence of the deflagration limit on binder type is due to differences in kinetics, flame temperature, or fuel molecule transport properties. A more general conclusion is instinctively drawn from the trend in limiting binder thickness with pressure, i.e., that the PVC flame is diffusion dominated down to 0.4 MPa. At lower pressure the binder thickness deflagration limit becomes extremely pressure dependent. This behavior suggests that the PVC flame has become kinetically limited. However, these speculations may be premature, in view of the intricate flame complex that is suggested by the present studies.

DISCUSSION AND SUMMARY

The objective of the present research is to determine the rate-controlling steps in combustion of heterogeneous propellants, starting with AP-HC propellants. The term "rate controlling" is used in a broad sense here, to mean steps that control various aspects of propellant combustion, such as burning rate, dynamic response to pressure and flow oscillations, and single transient events such as ignition, DDT, and quenching (the controlling steps are not the same in all of these aspects of combustion). The immediate goal is less ambitious, i.e., to determine the structure of the steady state combustion zone, and how it depends on pressure, ingredient particle size, etc. The sandwich burning approach was adopted in order to obtain less ambiguous test results and easier interpretation of those results.

Because of the microscopic, two-dimensional nature of the combustion zone, real-time observations by combustion photography and other methods provide only indirect information, and resort has been taken to "post mortem" observations, i.e., high resolution microscopic examination of samples from quenched burning experiments (augmented by burning rate and deflagration limit experiments). Determination of the combustion zone microstructure involves reconciliation of the results of a variety of such experiments, and known principles of combustion science to produce a combustion theory that will stand the test of subsequent experiments, or correctly predict their outcome. The results to date are controversial enough to raise discussion and pose new issues and experiments. They are well founded enough to encourage continued investigation with the expectation that a new and more realistic theoretical model will result. They are preliminary, and hence subject to some extension and "quantification". The picture of the combustion zone that has emerged is shown in Fig. 14. The principal features are:

1. An exothermally reacting froth on most of the AP surface. The froth appears to be present everywhere that the multiple flame complex maintains a surface temperature as high as that present during normal AP self-deflagration.
2. An AP flame, close to the AP surface, present whenever the froth layer is present.

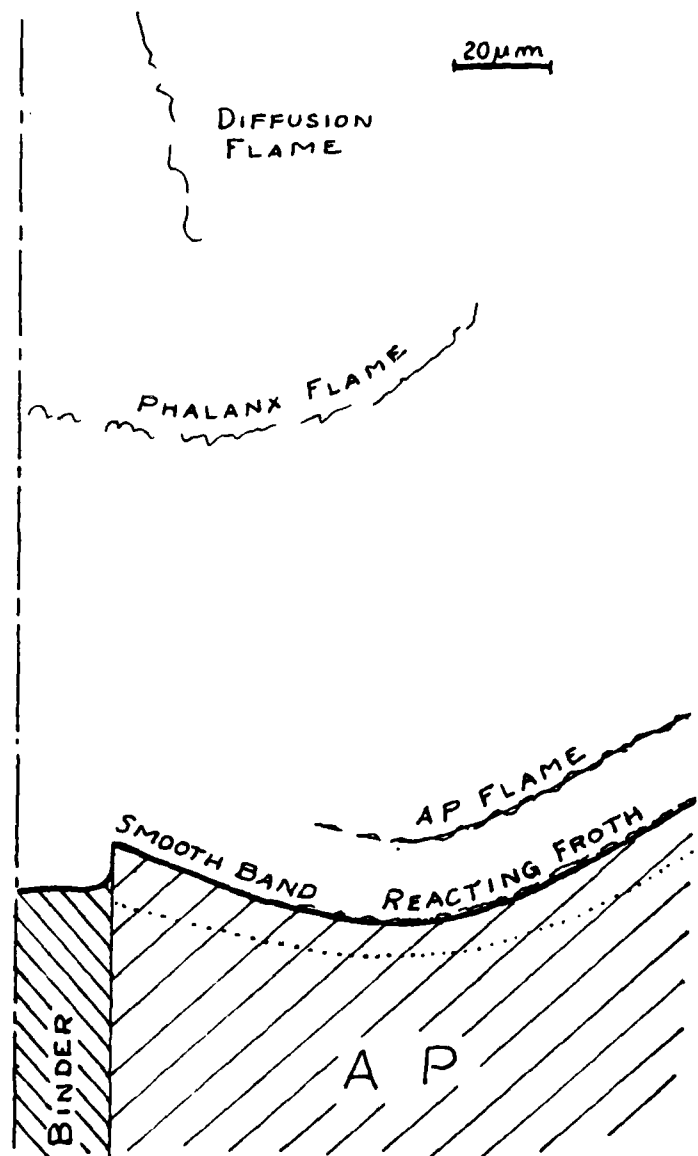


Fig. 14 Postulated combustion zone microstructure for thin-binder sandwiches.

3. A "broad", kinetically limited "phalanx" flame, more remote from the surface than the AP flame, but still affecting surface regression for some distance out from the AP-binder interface plane.
4. A diffusion limited flame still more remote from the leading edge of the burning front (one for each AP-binder interface plane?), absent when the binder is so thin that the fuel is all consumed in the phalanx flame.
5. A "smooth band" region on the AP surface adjoining the edge of the AP-binder interface plane; the reacting froth, and probably the AP flame are absent here, and AP vapors (i.e., NH_3 , HClO_4 , and other intermediate products) mix with fuel vapors and feed the phalanx flame.
6. A resulting surface profile with recessed binder (when not too thick), a feature that reflects the lower temperature required for decomposition of binder at the rates present. The AP regression is retarded in the smooth band region because of heat loss to the binder and resultant local quenching of AP exothermic reactions. The leading edge of the AP burning front is outside the smooth band, where the AP flame is less impeded by heat drain and supported by heat transfer from the phalanx flame.
7. Combustion zone dimensions are roughly as shown in Figure 14 typical of a pressure of 5 MPa.

At low pressure the phalanx flame is presumably further from the burning surface, and the AP flame may fail to be sustained by the phalanx flame. With sufficiently thin binder, the phalanx flame may consume all of the fuel (no diffusion flame). At high pressure (> 7 MPa), the AP self-deflagration rate is higher than the rate of the overall flame complex, so the region near the interface protrudes. The details of the flame complex, consequently, become less important to burning rate, but are probably still important to other aspects of combustion (e.g., in propellant combustion). The faster kinetics at higher pressure presumably cause the phalanx flame to move in closer to the surface, in the process limiting the amount of pre-mixed reactants available and increasing the contribution of the diffusion limited part of the flame. At high pressure, the net effect is a retardation of burning rate relative to the AP rate, so that the AP portion of the sandwich dominates burning rate.

In summary, the principal results of recent work are:

1. Further detailed verification of previously reported quench profiles vs pressure, binder thickness, and kind of binder. Particular emphasis was placed on verifying the details of smooth bands and retarded regression rates near the AP-binder interfaces. Efforts are now directed at obtaining quantitative data on these details of combustion zone microstructure.
2. Exploration of the effect of changes in "fuel" lamina material to clarify the mechanistic basis of smooth bands and retardation of regression rate (mica, gold, and AP-AP without any intervening lamina). This investigation is continuing.
3. Preliminary efforts at determination of burning rates of thin-binder sandwiches in the < 5 MPa domain where current methods and data are inadequate. The purpose of these experiments is to establish the curves in Fig. 13 on a factual basis. Efforts are continuing.
4. Systematic determination of the sandwich deflagration limit (binder thickness vs pressure) for three binders.
5. Development of a qualitative model for sandwich burning, as a preliminary to quantitative modeling.

REFERENCES

1. Price, E. W., J. C. Handley, W. C. Strahle, T. S. Sheshadri, R. K. Sigman and Ashoke Ghosh, "Combustion Mechanisms of Solid Propellants," Final Technical Report for Office of Naval Research, Contract N00014-75-C-0332, Georgia Institute of Technology, September 1980.
2. Price, E. W., J. C. Handley, R. R. Panyam, R. K. Sigman and A. Ghosh, "Combustion of Ammonium Perchlorate-Polymer Sandwiches," AIAA Journal, Vol. 19, No. 3, 19, 1981, pp. 380-386.
3. Hightower, J. D., and E. W. Price, "Experimental Studies Relating to the Combustion Mechanism of Composite Propellants," Astronautica Acta, Vol. 14, 1968, pp. 11-21.
4. Boggs, T. L., D. E. Zurn, W. C. Strahle, J. C. Handley and T. T. Milkie, "Mechanisms of Combustion," Naval Weapons Center, NWC TP 5514, 1973.
5. Boggs, T. L., and D. E. Zurn, "The Deflagration of Ammonium Perchlorate-Polymeric Binder Sandwich Models," Combustion Science and Technology, Vol. 4, 1972, pp. 279-292.
6. Price, E. W., R. R. Panyam and R. K. Sigman, "Microstructure of the Combustion Zone: Thin-Binder AP-Polymer Sandwiches," 17th JANNAF Combustion Meeting, Sept. 22-26, 1980, CPIA Publication 329, Vol. I, November 1980, pp. 37-51.
7. Varney, A. M., and W. C. Strahle, "Experimental Combustion Studies of Two-Dimensional Ammonium Perchlorate-Binder Sandwiches," Combustion Science and Technology, Vol. 4, 1972, pp. 197-208.
8. Price, E. W., R. K. Sigman and J. C. Handley, "Microstructure of the Combustion Zone of Composite Solid Propellants," 15th JANNAF Combustion Meeting, September 1978, CPIA Publication 297, Vol. II, February 1979.
9. Fenn, J. B., "A Phalanx Flame Model for the Combustion of Composite Solid Propellants," Combustion and Flame, Vol. 12, June 1968, pp. 201-216.
10. Beckstead, M. W., R. L. Derr and C. F. Price, "A Model of Solid Propellant Combustion Based on Multiple Flames," AIAA Journal Vol. 8, No. 12, 1970.
11. Summerfield, M., G. S. Sutherland, M. J. Webb, H. J. Tabak and K. P. Hall, "Burning Mechanisms of Ammonium Perchlorate Propellants," Progress in Astronautics and Rocketry, Vol. 1, Solid Propellant Rocket Research, Academic Press, New York, 1960, pp. 141-182.

12. Martynyuk, V. F., N. N. Bakhman and I. N. Lobanov, "Catalysis and Inhibition of Combustion in an Ammonium Perchlorate-Polymethyl Methacrylic Laminate System," Combustion, Explosion, and Shock Waves, Vol. 13, No. 2, March-April 1977, pp. 147-155.
13. Strahle, W. C., "Solid Propellant Sandwich Deflagration Analysis," AIAA Journal, Vol. 13, No. 5, May 1975, pp. 640-646.
14. Bakhman, N. N., and V. B. Librovich, "Flame Propagation Along Solid Fuel-Solid Oxidizer Interface," Combustion and Flame, Vol. 15, 1970, pp. 143-153.
15. Ermolaev, B. S., A. I. Korotkov and Yu. V. Frolov, "Laws of Combustion of a Solid-Propellant Sandwich," Combustion, Explosion, and Shock Waves, Vol. 6, No. 3, July-Sept., 1970, pp. 277-285.
16. Ermolaev, B. S., A. I. Korotkov and Yu. V. Frolov, "Study of the Action of Catalysts Using Layered Systems," Combustion, Explosion, and Shock Waves, Vol. 5, No. 2, 1969, pp. 286-289.
17. Belyaev, A. F., and N. N. Bakhman, "Theory of Burning of Powders and Solid Rocket Propellants (Review)," Combustion, Explosion, and Shock Waves, Vol. 2, No. 4, 1966, pp. 1-10.
18. Boggs, T. L., E. W. Price and D. E. Zurn, "The Deflagration of Pure and Isomorphously Doped Ammonium Perchlorate," 13th Symposium (International) on Combustion, 1971, pp. 995-1008.
19. Cohen Nir, Ephraim, "An Experimental Study of the Low Pressure Limit for Steady Deflagration of Ammonium Perchlorate," Combustion and Flame, Vol. 20, 1973, pp. 419-435.
20. Hightower, J. D., E. W. Price, and D. E. Zurn, "Continuing Studies of the Combustion of Ammonium Perchlorate," CPIA Publication No. 162, Vol. 1, December 1967, p. 527-534.

DISTRIBUTION LIST

No. Copies	No. Copies	No. Copies	No. Copies
<p>Assistant Secretary of the Navy (R, E, and S) Room 5E 73 Pentagon Washington, DC 20350 Attn: Dr. L. V. Schmidt</p> <p>Scientific Advisor Commandant of the Marine Corps Code RD-1 Washington, DC 20380 Attn: Dr. A. L. Safflosky</p> <p>Office of Naval Research Code 413 Arlington, VA 22217 Attn: Dr. Richard S. Miller</p> <p>Office of Naval Research Code 260 Arlington, VA 22217 Attn: Mr. David Siegel</p> <p>Office of Naval Research Western Office 1030 East Green Street Pasadena, CA 91106 Attn: Dr. R. J. Marcus</p> <p>Office of Naval Research East Central Regional Office 666 Summer Street, Bldg. 114-D Boston, MA 02210 Attn: Dr. Larry Peablies</p> <p>Office of Naval Research San Francisco Area Office One Hallide Plaza, Suite 601 San Francisco, CA 94102 Attn: Dr. Phillip A. Miller</p> <p>Aerojet Solid Propulsion Company P. O. Box 13400, Bldg. 208/Dept. 4350 Sacramento, CA 95813 Attn: Mr. Michael J. Ditore</p> <p>Aerojet Strategic Propulsion Co. P. O. Box 13459C Sacramento, CA 95813 Attn: Dr. R. L. Lou</p>	<p>Aerospace Corporation P. O. Box 92917 Los Angeles, CA 90043 Attn: Ellis M. Landsbaum</p> <p>U. S. Air Force Academy F35RL/NC USAF Academy, CO 80840 Attn: Dr. John S. Wilkes, Jr.</p> <p>AFATL Code DLDL Eglin AFB, FL 32542 Attn: Mr. Otto K. Helney</p> <p>Air Force Office of Scientific Research Directorate of Aerospace Sciences Bolling Air Force Base Washington, DC 20332 Attn: Dr. L. H. Caveny</p> <p>Air Force Office of Scientific Research Directorate of Chemical Sciences Bolling Air Force Base Washington, DC 20332 Attn: Mr. Donald L. Ball</p> <p>AFRPL Code PACC Edwards AFB, CA 93523 Attn: Mr. W. C. Andrepoint</p> <p>AFRPL Code MKP/MS24 Edwards AFB, CA 93523 Attn: Mr. R. Geisler</p> <p>AFRPL Code DYSC Edwards AFB, CA 93523 Attn: Mr. Daweel George</p> <p>AFRPL Code LK Edwards AFB, CA 93523 Attn: LTC B. Loving</p>	<p>AFRPL Code MKPA Edwards AFB, CA 93523 Attn: Dr. F. Roberto</p> <p>AFRPL Code CA Edwards AFB, CA 93523 Attn: Dr. R. R. Weiss</p> <p>Anal-Syn Lab Inc. P. O. Box 347 Paoli, PA 19301 Attn: Dr. V. J. Keenan</p> <p>Army Ballistic Research Labs ARRADCOM Code DRDAR-BLI Aberdeen Proving Ground, MD 21005 Attn: Mr. J. M. Frankie Dr. Ingo W. May Mr. L. A. Watermeier</p> <p>Army Ballistic Research Labs ARRADCOM Code DRDAR-BLP Aberdeen Proving Ground, MD 21005 Attn: Dr. A. W. Barrows</p> <p>Army Ballistic Research Labs ARRADCOM Code DRDAR-RLT Aberdeen Proving Ground, MD 21005 Attn: Dr. Philip Howe</p> <p>Army Frankford Arsenal Bridge & Tacony Streets Philadelphia, PA 19137 Attn: J. Lannon</p> <p>HQ US Army Material Development Readiness Command Code DRCDL-DW 501 Eisenhower Avenue Room 8N92 Alexandria, VA 22333 Attn: Mr. S. R. Matos</p> <p>Army Missile Command Code DRSMI-R Redstone Arsenal, AL 35898 Attn: Dr. R. G. Rhoades</p>	<p>Army Missile Command Code DRSMI-RKL Redstone Arsenal, AL 35898 Attn: Dr. W. W. Wharton</p> <p>Army Research & Development Command ARRADCOM Code LCWSL Dover, NJ 07802 Attn: Mr. C. Lenchitz</p> <p>Army Research & Development Command ARRADCOM Code DRDAR-SCA-PE Dover, NJ 07802 Attn: Mr. L. Stiefel</p> <p>Army Research & Development Command ARRADCOM Code DRDAR-LCE Dover, NJ 07801 Attn: Dr. R. F. Walker</p> <p>U. S. Army Research Office Chemical & Biological Sciences Div. P. O. Box 1221 Research Triangle Park, NC 27709</p> <p>Atlantic Research Corp. 5390 Cherokee Ave. Alexandria, VA 22314 Attn: Dr. C. B. Henderson Dr. Merrill K. King</p> <p>Atlantic Research Corp Pine Ridge Plant 7511 Wellington Rd. Gainesville, VA 22063 Attn: Dr. W. D. Stephens</p> <p>Brigham Young University Provo, UT 84601 Attn: Dr. Merrill W. Beckstead</p> <p>British Embassy Munitions Directorate Propellants and Explosives Defence Equipment Staff 3100 Massachusetts Ave. Washington, DC 20008 Attn: Dr. T. Sinden</p>

DISTRIBUTION LIST

No. Copies	No. Copies	No. Copies	No. Copies
California Institute of Tech. Graduate Aeronautical Lab Pasadena, CA 91125 Attn: Prof. W. G. Knauss	1	Georgia Institute of Tech. School of Aerospace Engineering Atlanta, GA 30332 Attn: Prof. Edward Price	1
California Institute of Tech. Dept. of Chemical Engineering Pasadena, CA 91125 Attn: Prof. N. W. Tschoegl	1	Hercules Inc. Aerospace Division Allegheny Ballistic Lab P. O. Box 210 Washington, DC 21502 Attn: Dr. Rocco C. Musso Dr. R. R. Miller	2
California Institute of Tech. 204 Karman Lab 1201 E. California St. Pasadena, CA 91109 Attn: Fred E. C. Culick	1	Hercules Inc. Beechus Works P. O. Box 98 Magna, UT 84044 Attn: Dr. E. H. Debutts Dr. James H. Thacher Dr. K. McCarty	3
California Institute of Tech. Jet propulsion Laboratory 4800 Oak Grove Drive Pasadena, CA 91103 Attn: Leon D. Strand	1	Hercules Inc. Eglin Code AFATL/DLIDL Eglin AFB, FL 32542 Attn: Dr. Ronald L. Simmons	1
Calspan Corporation P. O. Box 235 Buffalo, NY 14221 Attn: Edward B. Fisher	1	Hercules Inc. Aerospace Division P. O. Box 210 Cumberland, MD 21502 Attn: Dr. Kenneth O. Hartman	1
Catholic Univ. of America Physics Department 520 Michigan Ave., NE Washington, DC 20017 Attn: Prof. T. Litovitz	1	IBM Research Lab D42.282 San Jose, CA 95193 Attn: Dr. Thor L. Smith	1
Mr. Norman Cohen 898-A Pine Ave. Redlands, CA 92373	1	Institute for Defense Analyses 400 Army-Navy Drive Arlington, VA 22202 Attn: R. C. Oliver	1
Cornell University School of Chemical Engineering Olin Hall Ithaca, NY 14853 Attn: Prof. F. Rodriguez	1	Johns Hopkins University APL Chemical Propulsion Info. Agency Johns Hopkins Road Laurel, MD 20810 Attn: Mr. Thomas W. Christian Mr. Theodore M. Gilliland	2
Defense Technical Information Center DTIC-DDA-2 Communication Station Alexandria, VA 22304	12		
Lawrence Livermore Laboratory University of California Code L-324 Livermore, CA 94550 Attn: Dr. R. McGuire	1	Lockheed Missile & Space Co. Code 83-10, Bldg. 154 P. O. Box 504 Sunnyvale, CA 94088 Attn: Dr. Jack Linsk	1
NASA/George C. Marshall Space Flight Center Code EP 24 Huntsville, AL 35812 Attn: Robert J. Richmond	1	NASA/HQ Code RP 600 Independence Ave., SW, Rm. 625 Washington, DC 20546 Attn: Frank W. Stephenson, Jr.	1
Naval Air Systems Command Code 330 Washington, DC 20361 Attn: Mr. R. Brown	1	Naval Air Systems Command Code AIR-310C Washington, DC 20360 Attn: Dr. H. Rosenwasser	1
Naval Air Systems Command Code 03P25 Washington, DC 20360 Attn: Mr. B. Sobers	1	Naval Air Systems Command Code NAIIR-93A-Tech Library Washington, DC 20361	1
Naval Explosives Dev. Engineering Department Assistant Director Naval Weapons Station Yorktown, VA 23691 Attn: Dr. L. R. Rothstein	1	Naval Explosive Ordnance Disposal Tech Center Code D Indian Head, MD 20640 Attn: Dr. Lionel Dickinson	1
Naval Materiel Command Strategic Systems Project Office Department of the Navy Room 901 Washington, DC 20376 Attn: Dr. J. F. Kincaid	1		

DISTRIBUTION LIST

No. Copies	No. Copies	No. Copies	No. Copies
<p>Naval Materiel Command Strategic Systems Project Office Propulsion Unit Code SP 273 Department of the Navy Washington, DC 20376</p> <p>Naval Materiel Command Strategic Systems Project Office Department of the Navy Room 1048 Washington, DC 20376 Attn: Mr. E. L. Throckmorton</p> <p>Naval Ocean Systems Center San Diego, CA 92132 Attn: Mr. Joe McCartney</p> <p>Naval Ocean Systems Center Marine Sciences Division San Diego, CA 92132 Attn: Dr. S. Yamamoto</p> <p>Naval Ordnance Station CM PM4 Indian Head, MD 20640 Attn: Mr. C. L. Adams</p> <p>Naval Ordnance Station Code 5253 Indian Head, MD 20640 Attn: Mr. S. Mitchell</p> <p>Naval Ordnance Station Indian Head, MD 20640 Attn: Mr. Peter L. Stang</p> <p>Naval Postgraduate School Dean of Research Monterey, CA 93940 Attn: Dr. William Tolles</p> <p>Naval Postgraduate School Physics & Chemistry Dept. Monterey, CA 93940 Attn: Prof. Richard A. Reinhardt</p> <p>Naval Postgraduate School Department of Aeronautics Monterey, CA 93940 Attn: Mr. David W. Netzer</p>	<p>Office of Naval Research Mechanics Program Code 432 Arlington, VA 22217 Attn: Dr. N. L. Rasdekas</p> <p>Naval Research Lab. Code 6100 Washington, DC 20375</p> <p>Naval Research Lab. Code 6310 Washington, DC 20375 Attn: Dr. J. Schur</p> <p>Naval Sea Systems Command Code SEA 64E Washington, DC 20362 Attn: Mr. R. Beauregard</p> <p>Naval Sea Systems Command Code 62R3 Washington, DC 20362 Attn: Mr. G. Edwards</p> <p>Naval Sea Systems Command Code 62R2 Washington, DC 20362 Attn: Mr. J. Murrin</p> <p>Naval Ship Engineering Center Materials Branch Philadelphia, PA 19112 Attn: Mr. John Boyle</p> <p>Naval Ship Research & Development Center Applied Chemistry Division Annapolis, MD 21401 Attn: Dr. G. Bosmajian</p> <p>Naval Surface Weapons Center Commander Silver Spring, MD 20910 Attn: Mr. G. B. Wilmot</p> <p>Naval Surface Weapons Center Code ROs White Oak Laboratory Silver Spring, MD 20910 Attn: Dr. D. J. Pastine</p>	<p>Naval Surface Weapons Center Code R10 White Oak Laboratory Silver Spring, MD 20910 Attn: Dr. S. J. Jacobs</p> <p>Naval Surface Weapons Center Code R11 White Oak Laboratory Silver Spring, MD 20910 Attn: Dr. H. G. Adolph Dr. T. Hall Dr. M. J. Kamlet Dr. K. F. Mueller</p> <p>Naval Surface Weapons Center Code R13 White Oak Laboratory Silver Spring, MD 20910 Attn: Dr. E. Zimmet Dr. R. Bernecker</p> <p>Naval Surface Weapons Center Code R16 Indian Head, MD 20640 Attn: Dr. T. D. Austin</p> <p>Naval Surface Weapons Center Code R101 Indian Head, MD 20640 Attn: Mr. G. L. Machenzle</p> <p>Naval Surface Weapons Center Code R121 White Oak Laboratory Silver Spring, MD 20910 Attn: Mr. M. Stosz</p> <p>Naval Surface Weapons Center Code R122 White Oak Laboratory Silver Spring, MD 20910 Attn: Mr. L. Roslund Chief MAT Code 0716 Washington, DC 20360 Attn: Dr. A. Faulstich</p> <p>Office of Naval Technology Chief MAT Code 0712 Washington, DC 20360 Attn: LCDR J. Walker</p>	<p>Naval Underwater Systems Center Energy Conversion Dept. Code 3B311 Newport, RI 02840 Attn: Mr. Robert S. Lazar</p> <p>Naval Weapons Center Code 385 China Lake, CA 93555 Attn: Dr. A. Amster Dr. A. Nielsen</p> <p>Naval Weapons Center Code 388 China Lake, CA 93555 Attn: Mr. T. L. Boggs Dr. R. L. Derr Dr. R. Reed, Jr.</p> <p>Naval Weapons Center Code 3205 China Lake, CA 93555 Attn: Mr. Lee N. Gilbert Dr. L. Smith Dr. C. Thelen</p> <p>Naval Weapons Center Code 3272 China Lake, CA 93555 Attn: Mr. R. McCarten</p> <p>Naval Weapons Center Code 3858 China Lake, CA 93555 Attn: Dr. E. Martin</p> <p>Naval Weapons Support Center Code 5042 Crane, IN 47522 Attn: Dr. B. Douda</p> <p>Northwestern University Dept. of Civil Engineering Evanston, IL 60208 Attn: Prof. J. D. Achenbach</p> <p>Pennsylvania State University Dept. of Mechanical Engineering University Park, PA 16802 Attn: Prof. Kenneth Kuo</p>

DISTRIBUTION LIST

No. Copies	No. Copies	No. Copies	No. Copies	No. Copies	
Princeton Combustion Research Laboratories, Inc. 1004 U. S. Highway One North Princeton, NJ 08540 Attn: Dr. Martin Summerfield	1	Southwest Research Institute Institute Scientist P. O. Drawer 28310 San Antonio, TX 78228 Attn: Mr. William H. McLain	1	United Technologies Research Center 455 20 400 Main Street East Hartford, CT 06108 Attn: Mr. R. H. W. Waesche	1
Princeton University School of Engineering and Applied Sciences Dept. of Mech. Eng. & Aero. Eng. The Engineering Quadrangle Princeton, NJ 08540 Attn: Dr. Forman A. Williams	1	SRI International 333 Ravenswood Ave. Menlo Park, CA 94025 Attn: Mr. M. Hill Dr. Y. M. Gupta	2	University of Akron Institute of Polymer Science Akron, OH 44325 Attn: Prof. Alan N. Gent	1
Purdue University School of Mechanical Engineering TSPC Chaffee Hall West Lafayette, IN 47906 Attn: Mr. John R. Osborn	1	Texas A & M University Dept. of Civil Engineering College Station, TX 77843 Attn: Prof. Richard A. Schapery	1	University of California Berkeley, CA 94720 Attn: Prof. A. G. Evans	1
Rockwell International Corp. Rocketdyne Division BA08 6633 Canoga Ave. Canoga Park, CA 91304 Attn: Mr. Joseph E. Flanagan	1	Thiokol Corporation Elkton Division P. O. Box 241 Elkton, MD 21921 Attn: Mr. E. S. Sutton Dr. C. W. Vriesen	2	University of California Dept. of Chemistry 905 Hilgard Ave. Los Angeles, CA 90024 Attn: Prof. M. D. Nicol	1
Rockwell International Corp. Rocketdyne Division P. O. Box 548 McGregor, TX 76637 Attn: Mr. William G. Haymes	1	Thiokol Corporation Government Systems Group Technical Director P. O. Box 9238 Ogden, Utah 84409 Attn: Dr. T. F. Davidson	1	University of Delaware Department of Chemistry Newark, DE 19711 Attn: Dr. T. C. Brill	1
Rohm and Haas Company Huntsville, AL 35801 Attn: Dr. H. Shuey	1	Thiokol Corporation Huntsville Division Huntsville, AL 35807 Attn: Dr. D. A. Flanagan Mr. G. F. Mangum Mr. J. D. Byrd	3	University of Illinois AE Dept. Transportation Building, Room 105 Urbana, IL 61801 Attn: Dr. Herman Krier	1
Sandia Laboratories Division 2913 P. O. Box 5800 Albuquerque, NM 87185 Attn: Dr. S. Sheffield	1	Thiokol Corporation Wasatch Division P. O. Box 324 Brigham City, UT 84302 Attn: Dr. J. C. Hingshaw Mr. John A. Peterson Dr. G. Thompson	3	University of Maryland Dept. of Mechanical Engineering College Park, MD 20742 Attn: Prof. R. W. Armstrong	1
Science Applications, Inc. Suite 423 1315 Ventura Blvd. Woodland Hills, CA 91364 Attn: Mr. R. B. Edelman	1	United Technologies Corp. Chemical Systems Division P. O. Box 338 Sunnyvale, CA 94088 Attn: Dr. Robert S. Brown Dr. C. M. Frey	2	University of Southern California Mechanical Engineering Dept. OHE 200 Los Angeles, CA 90007 Attn: Dr. M. Gerstein	1
Space Sciences, Inc. 135 Maple Avenue Monrovia, CA 91016 Attn: Dr. M. Farber	1			University of Utah Salt Lake City, UT 84112 Attn: Dr. G. A. Flandro	1
				University of Utah Dept. of Mech. & Ind. Eng. MEB 3008 Salt Lake City, UT 84112 Attn: Dr. Stephen Swanson	1
				University of Waterloo Dept. of Mechanical Engineering Waterloo, Ontario CANADA Attn: Dr. Clarke E. Hermance	2
				Washington State University Dept. of Physics Pullman, WA 99163 Attn: Prof. G. D. Duwall Prof. T. Dickinson	1
				Whittaker Corporation Bermite Division 2216 W. Soledad Canyon Road Saugus, CA 90024 Attn: Mr. L. Bloom	1